(The actual exam will have 15 questions.)

1. The density of the liquid flowing through the horizontal pipe in the drawing is 1200 kg/m^3 . The radius of the pipe at point A is 0.50 m and the radius of the pipe at point B is 0.25 m. If the flow speed at point A is 7 m/s, what is the difference in pressure, $P_{\rm B} - P_{\rm A}$, between points B and A?



- 2. The coefficient of linear expansion of steel is $12 \times 10^{-6}/C^{\circ}$. A railroad track is made of individual rails of steel 1.0 km in length. By what length would these rails change between a cold day when the temperature is 14 °F and a hot day at 86 °F? B) 24 cm A) 0.62 cm C) 48 cm D) 480 cm E) 620 cm
- 3. A thermos bottle contains 3.0 kg of water and 2.0 kg of ice in thermal equilibrium at 0° C. How much heat is required to bring the system to thermal equilibrium at 50 °C? A) 1.05×10^6 J B) 1.30×10^6 J C) 1.72×10^6 J D) 2.26×10^6 J E) $1.13 \times 10^7 \text{ J}$
- 4. Complete the following statement: The space between the inner walls of a thermos bottle is evacuated to minimize heat transfer by
 - A) radiation. B) conduction.
 - D) conduction and radiation.
- C) conduction, convection, and radiation. E) conduction and convection.
- 5. A slab of insulation is made of three layers, as Drawing I indicates. Each of the layers A, B, and C has the same thickness, but a different thermal conductivity. Heat flows through the slab, and the temperatures are as shown. What are the temperatures T_1 and T_2 in Drawing II where the layers are arranged in a different order?
 - A) $T_1 = 230 \text{ °C}$ and $T_2 = 170 \text{ °C}$ B) $T_1 = 200 \text{ °C}$ and $T_2 = 180 \text{ °C}$ C) $T_1 = 220 \text{ °C}$ and $T_2 = 160 \text{ °C}$ D) $T_1 = 180 \text{ °C}$ and $T_2 = 160 \text{ °C}$
 - E) $T_1 = 210 \text{ °C}$ and $T_2 = 190 \text{ °C}$
- 240 °C 210 °C 150 °C 250 °C С Α В Drawing I 2 150 °C 250 °C В С A Drawing II

- 6. How many moles are in a 0.53-kg sample of sulphur dioxide, SO₂? (atomic masses: C = 32 u; O = 16 u)
 - A) 5.2
 - B) 8.3
 - C) 48
 - D) 1.6×10^4
 - E) 5.0×10^{24}
- 7. At what temperature would one mole of molecular nitrogen (N₂) have 7.0 × 10³ J of *translational* kinetic energy? Note: the atomic mass of N is 14 u.
 A) 130 °C B) 290 °C C) 480 °C D) 560 °C E) 720 °C
- 8. Enclosed beneath the moveable piston in the drawing is 4.8 moles of a monatomic ideal gas. The gas performs work on the piston as 2300 J of heat are added from the surroundings. During the process, the temperature of the gas decreases by 45 K. How much work does the gas perform?



9. An ideal monatomic gas undergoes an adiabatic process; and its internal energy *decreases* by 50 J. Which pair of choices below is correct for this process?

work done	heat exchanged
A) 50 J by the system	zero joules
B) 50 J on the system	zero joules
C) 50 J by the system	100 J supplied
D) zero joules	50 J removed
E) zero joules	50 J added

- 10. A heat engine operates in a Carnot cycle between reservoirs of temperatures 1000 K and 400 K. It is found to discharge 20 J of heat per cycle to the cold reservoir. What is the work output per cycle?
 - A) 10 J
 - B) 20 J
 - C) 30 J
 - D) 40 J
 - E) 50 J

11. The speed of sound in a certain metal block is 2.00×10^3 m/s. The graph shows the amplitude (in meters) of a wave traveling through the block versus time (in milliseconds). What is the wavelength of this wave?



12. A steel wire of mass 0.400 kg and length 0.640 m supports a 102-kg block. The wire is struck exactly at its midpoint causing a small displacement. How long does it take the peak of this displacement to reach the top of the wire?



- 13. The decibel level of a jackhammer is 125 dB relative to the threshold of hearing. Determine sound intensity produced by the jackhammer.
 - A) 1.0 W/m²
 B) 3.2 W/m²
 C) 4.8 W/m²
 D) 12 W/m²
 E) 88 W/m²

Answers

- 1. E
- 2. C
- 3. C
- 4. E
- 5. C
- 6. B
- 7. B
- 8. A
- 9. A
- 10. C
- 11. D
- 12. D
- 13. B