

## L 18 Thermodynamics [3]

- Heat transfer processes
  - convection
  - conduction
  - → radiation
- Physics of the atmosphere
  - the ozone layer
  - Greenhouse effect
  - climate change

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## Thermodynamics- review

- **Thermodynamics** is the science dealing with heat, work, and energy and the transformation of one into the other
  - **Heat** is disordered energy – random motion of molecules
  - **Work** is ordered or organized energy
- The laws of thermodynamics are a set of empirical (based on observations) rules that place limits on the transformations

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## Heat as moving light → radiation

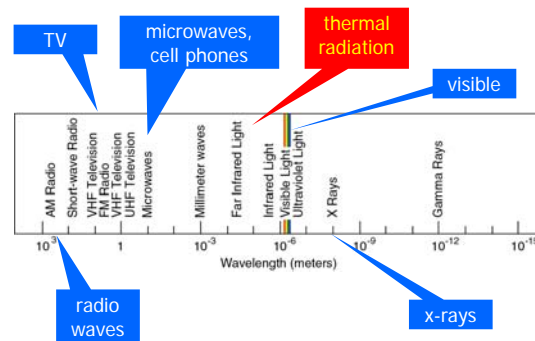


Radiation emitted by a heating element

- heat can be transferred by the emission of electromagnetic waves – thermal “light waves”, *invisible* to our eyes
- thermal radiation is a small part of the electromagnetic spectrum – called *infrared radiation*
- waves are characterized by their frequency or wavelength
- different colors in the visible correspond to different wavelengths from red to blue

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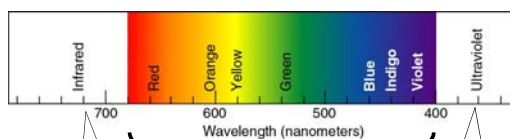
## electromagnetic spectrum



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## visible electromagnetic waves: LIGHT

shorter wavelength → more energy



thermal radiation

visible light

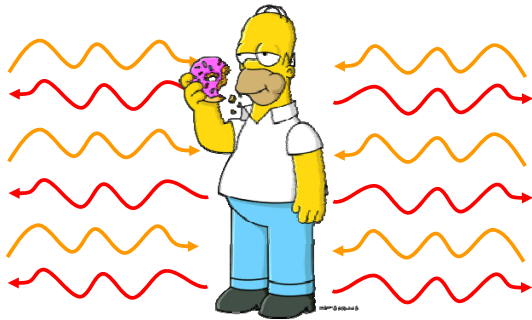
UV radiation produces sunburn

## What produces thermal radiation?

- All objects whose temperature is above absolute zero emit **thermal radiation**
- We continuously **emit** thermal radiation and **absorb** it from objects and people around us
- If we just emitted radiation we would eventually cool to absolute zero!
- The rate (J/s or Watts) at which thermal energy is radiated is given by  $P_{\text{radiation}} = \sigma e A T^4$ , where  $\sigma$  is a constant,  $A$  is the area of the object,  $T$  is its temperature in K, and  $e$  is a number between 0 and 1 called the emissivity (poor emitters have a small value of  $e$  and good emitters have  $e \approx 1$ ).

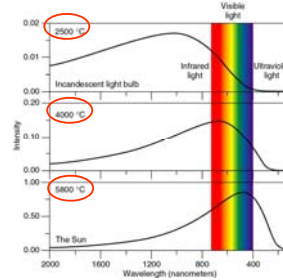
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## Emission and Absorption are balanced



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## Thermal radiation spectrum



- The intensity of radiation increases with temperature
- the color shifts toward the blue at higher temperatures
- The UV radiation from the sun is just beyond the violet (11,000 F)

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## sources of thermal radiation

- the incandescent light bulb (the ones that have a filament) are sources of both visible light and heat.
- when electricity flows through a wire it gets hot.
- it emits radiation even though you can't see it
- as it gets hotter it glows red then orange then white

tungsten filament, has a very high melting point, 3400 C



evacuated glass bulb

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## good emitters are good absorbers

- an object that is a good emitter is also a good absorber of thermal radiation
- a poor emitter is also a poor absorber
- generally dark, dull objects are the best emitters/absorbers
- shiny objects are poor emitters/absorbers, they are *good reflectors of radiation*
- If you do not want the edges of your pie to burn, you wrap it in aluminum foil. The aluminum foil reflects the heat rather than absorbing it.

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## good/bad emitters-Leslie's cube

copper cube filled with hot water



infrared radiation sensor

Even though all sides are at the same temperature, the black sides emit more radiation.

this side is painted black

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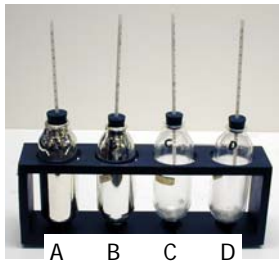
## Practical considerations

- wear light clothing in summer → light clothing absorbs less sunlight
- cover all body parts in winter → warm body parts (like your head) emit radiation → wear a hat

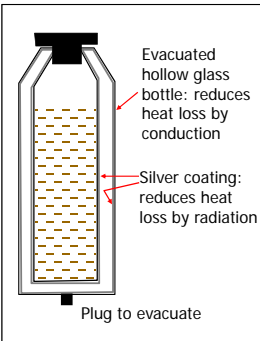


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## Which thermos bottle is best?



- A. silvered and evacuated
- B. silvered and un-evacuated
- C. evacuated
- D. un-silvered and un-evacuated



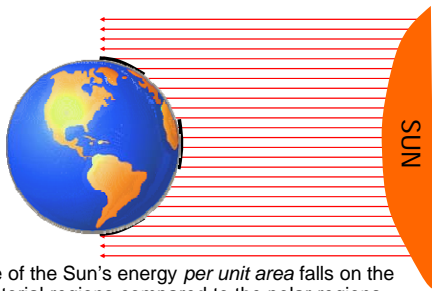
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## Physics of the atmosphere

- How the sun warms the earth
- The ozone layer issue
- Greenhouse effect
- **Climate change:** we share one planet with one atmosphere - the issues are global, and involve science, international politics, and economics

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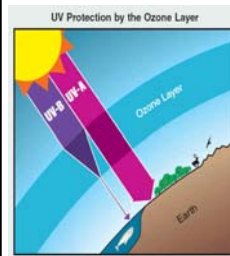
## Why is it colder at the poles than at the equator?



- More of the Sun's energy *per unit area* falls on the equatorial regions compared to the polar regions
- the earth reflects about 30% of incident solar energy
- **without the atmosphere the earth would be 30° C cooler!**
- Seasons are due to change in **tilt** of the earth

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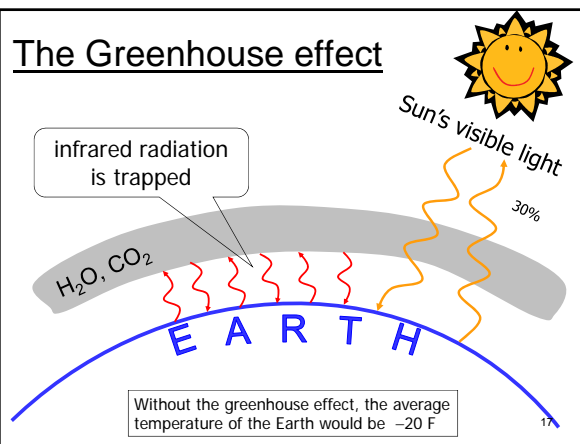
## The ozone layer: blocks UV-B rays



- ozone,  $O_3$  is a naturally occurring trace element in the atmosphere
- It absorbs solar ultraviolet radiation, especially the harmful UV-B rays
- Ozone is destroyed by CFC's (chlorofluorocarbons)
- loss affects us and environment
- Long-term observations reveal that Earth's ozone has been strengthening following international agreements to protect this vital layer of the atmosphere.

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## The Greenhouse effect



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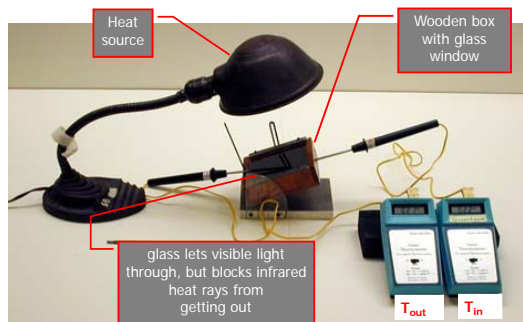
## Effect of greenhouse gases:

$H_2O$ ,  $CO_2$ ,  $CH_4$ , . . .

- the sun's visible light can penetrate through the atmosphere to the earth's surface and heat it
- the visible light energy is converted to thermal light energy
- the thermal radiation is reflected from the greenhouse gases in the atmosphere
- $CO_2$  concentrations are about 0.04% and increasing
- $CO_2$  produced by burning fossil fuels
- Water vapor accounts for up to 66%

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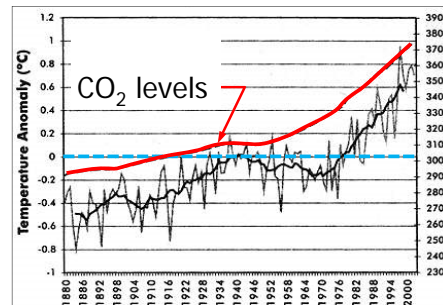
## Greenhouse effect Demo



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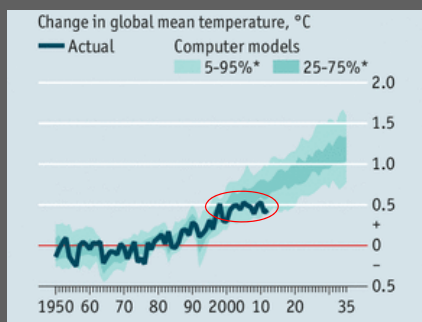
## Temperature change 1880-2003

the temperature anomaly is the difference between the current temperature and a long-term average value



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## No temperature rise over the last 15 years



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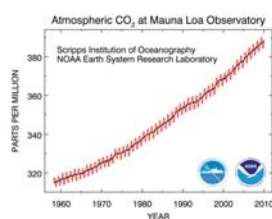
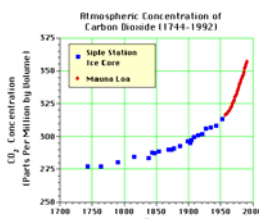
## What are climate *forcings*?

- Many factors affect the Earth's climate
- These factors are called *forcings* because they can drive or force the climate system to change
- Most important *forcings* during the last 1000 yrs.
  - changes in the output of energy from the sun
  - volcanic eruptions (injects dust into the atm.)
  - changes in the concentration of greenhouse gases in the atmosphere
- The big issues – are the changes:
  - natural or man-made (anthropogenic)
  - Self-reversible or require intervention

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## Greenhouse effect and climate change

- concentrations of CO<sub>2</sub> have been increasing
- → rise in earth's temperature
- similar effect occurs in your car during the day.



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## Climate change

- Are climate changes part of a natural cycle or driven by human activity (*anthropogenic*)?
- A recent statement signed by 256 members of US National Academy of Science (Science, 5/7/10)
  - There is always uncertainty associated with science, science never absolutely proves anything
  - Taking no action on climate change poses a dangerous risk for our planet
- Conclusions
  - The planet is warming due to increased concentrations of heat-trapping gases in our atmosphere
  - Most of the increase in the concentration over the last century is due to human activities, especially the burning of fossil fuels and deforestation (*controversial*)

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### Climate change, continued

- Natural causes also play a role but are now being overwhelmed by human-induced changes
- Warming the planet will cause climatic patterns to change at unprecedented speeds
- Policy makers should move forward to address the causes of climate change and reduce the threat of global climate change
- Effective actions are possible, but delay is not an option
- What are the social, political, and economic repercussions of taking or not taking action?

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