

Modern Physics (Phys. IV): 2704

Professor Jasper Halekas
Van Allen 70
MWF 12:30-1:20 Lecture

Final Exam Details

- Final Exam is Ready
 - 10 questions
 - ~1/3 points on material on first midterm
 - ~1/3 points on material on second midterm
 - ~1/3 points on material since second midterm
 - Sample final and extra questions on last 1/3 posted
- You are allowed a calculator, and both sides of an 8.5x11 page for an equation sheet
- Final 7:30-9:30 am Friday 5/11 in Van 70 (this room)
 - Donuts on me!

Special Relativity

- The laws of physics are the same in all inertial reference frames
- The speed of light is the same in all inertial frames

$$x' = \frac{x - vt}{\sqrt{1 - \frac{v^2}{c^2}}}$$

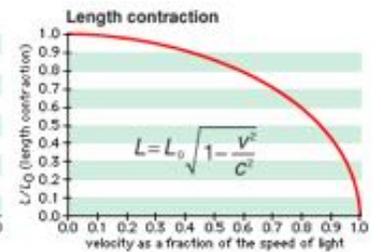
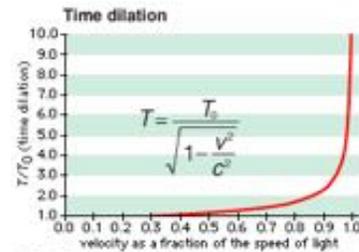
$$y' = y$$

$$z' = z$$

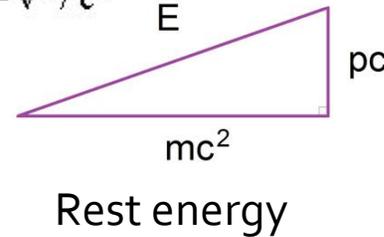
$$t' = \frac{t - \frac{vx}{c^2}}{\sqrt{1 - \frac{v^2}{c^2}}}$$



Time dilation
 Length contraction
 Velocity addition
 Doppler effect
 Relativity of simultaneity



$$E = \frac{mc^2}{\sqrt{1 - v^2/c^2}}$$



$$p = \frac{m_0 v}{\sqrt{1 - \frac{v^2}{c^2}}}$$

Sample Question

- A muon is produced in the upper atmosphere. At rest, the lifetime of a muon is 2.5 ns. However, we observe it traveling the entire 100 km to the Earth's surface (which at its velocity takes 250 ns according to our clock) without decaying. In the frame of the muon, how far does it travel?
 - A. 1 km
 - B. 10 km
 - C. 100 km
 - D. 1000 km
 - E. 10000 km

$$\Delta t = 2.5 \text{ ns}$$

$$\Delta t' = \gamma \Delta t = 250 \text{ ns}$$

$$\Rightarrow \gamma = 100$$

$$\Delta L = \Delta L' / \gamma = \boxed{1 \text{ km}}$$

Sample Question

- Two particles of mass m are moving at $v = 0.866c$ from opposite directions towards a fixed point. They collide head-on, resulting in the formation of a new particle of mass M at rest. What is M ? Note that at $v = 0.866c$, $\gamma = 2$.
- A. $M = m$
 - B. $M = 2m$
 - C. $M = 4m$
 - D. $M = 1.5m$
 - E. None of the above

$$E = E_1 + E_2 = \gamma_1 mc^2 + \gamma_2 mc^2 = 4mc^2$$

$$\vec{p} = \vec{p}_1 + \vec{p}_2 = 0$$

$$Mc^2 = \sqrt{E^2 - (pc)^2} = 4mc^2$$

$$\Rightarrow \boxed{M = 4m}$$

Particle Nature of EM Radiation

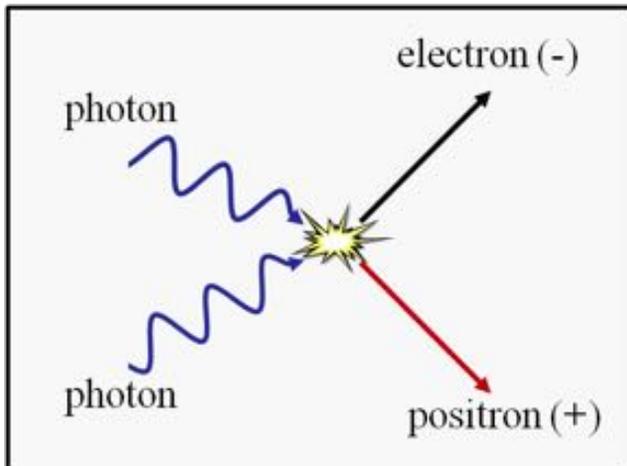
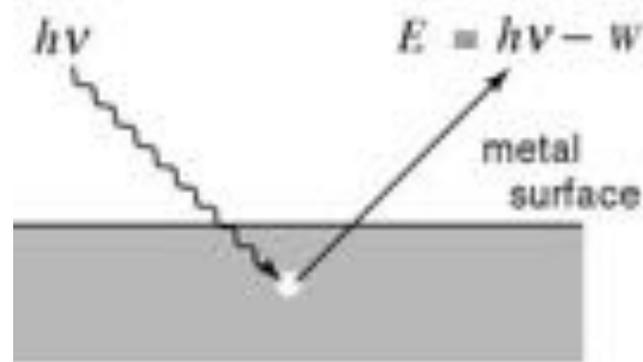
$$E = hf = \frac{hc}{\lambda} = \hbar\omega$$

$h = 6.63 \times 10^{-34} \text{ Js}$ → Planck constant

f = frequency of photon/electromagnetic radiation

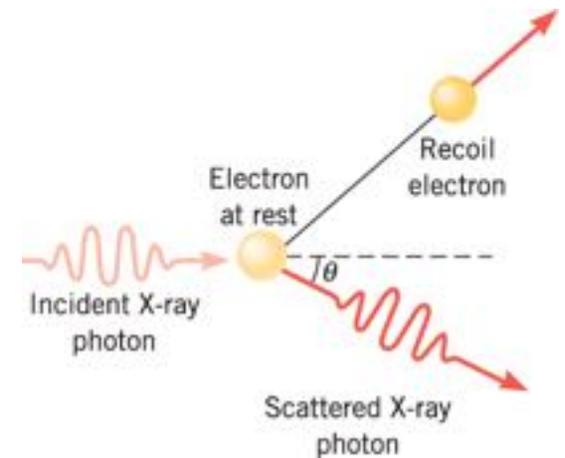
$c = 3 \times 10^8 \text{ m/s}$ → speed of light in a vacuum

λ = wavelength of photon/electromagnetic radiation



Collisions must satisfy conservation of (relativistic) energy and linear momentum

E.g. pair production, which can occur if the photon energy is more than $2 \times 511 \text{ keV}$ ($m_e c^2 = 511 \text{ keV}$)

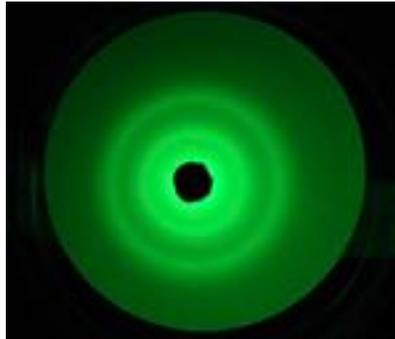
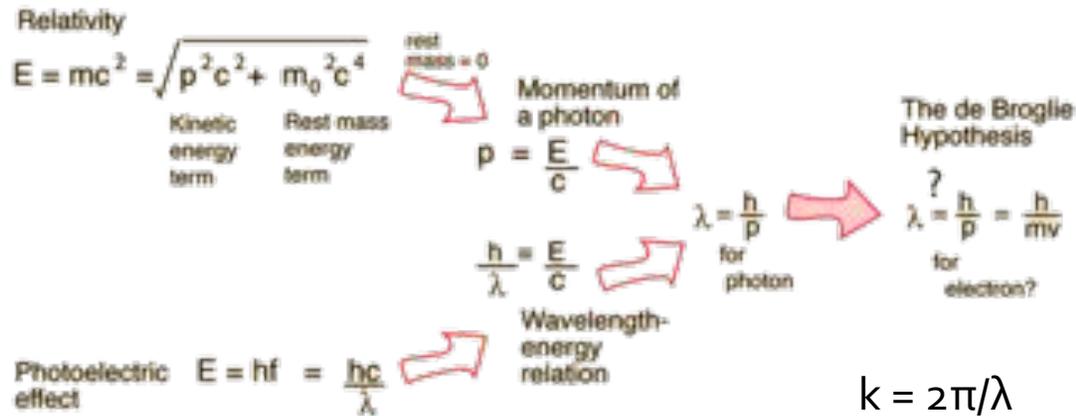


Sample Question

- An electron and a positron with total energies E_+ and E_- collide and produce two photons with wavelengths λ_1 and λ_2 . What is the correct conservation of energy equation?
 - A. $E_+ + E_- = hc/\lambda_1 + hc/\lambda_2$
 - B. $KE_+ + KE_- = hc/\lambda_1 + hc/\lambda_2$
 - C. $E_+ + E_- + m_e c^2 + m_e c^2 = hc/\lambda_1 + hc/\lambda_2$
 - D. $E_+ + E_- = h\lambda_1 + h\lambda_2$
 - E. $KE_+ + KE_- = h/\lambda_1 + h/\lambda_2$

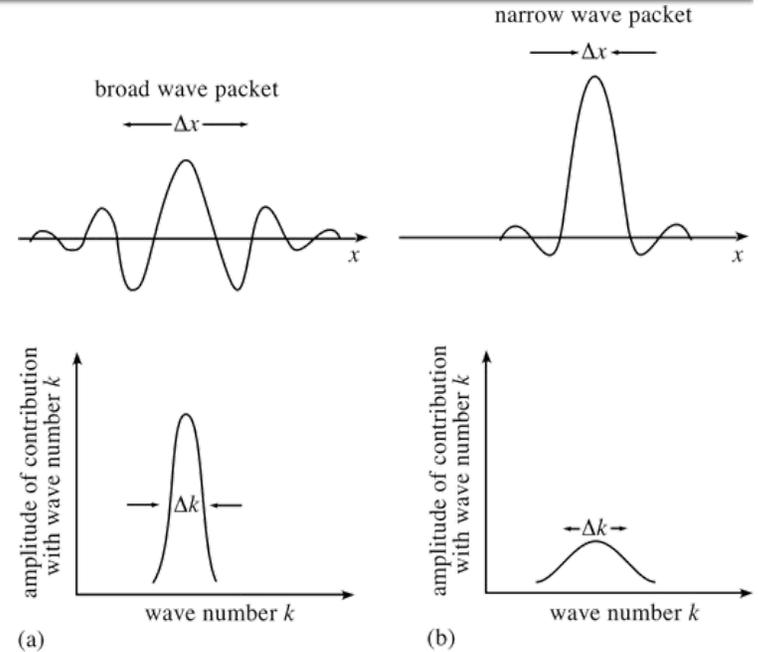
$$\begin{aligned} E_+ + E_- &= h\nu_1 + h\nu_2 \\ &= \frac{hc}{\lambda_1} + \frac{hc}{\lambda_2} \end{aligned}$$

Wavelike Properties of Particles



$$k = 2\pi/\lambda$$

$$k = p/\hbar$$



$$\Delta p \Delta x \geq \frac{1}{2} \hbar$$

$$\Delta E \Delta t \geq \frac{1}{2} \hbar$$

Sample Question

- A beam of electrons is incident on a slit, producing an interference pattern of maxima and minima. The slit width is halved. What happens to the spacing of the maxima/minima?
 - A. They get closer
 - B. They get farther apart
 - C. They stay the same

$$\Delta p_y \Delta y \sim \hbar$$

$$\Rightarrow \Delta p_y \sim \hbar / \Delta y$$

$$\Delta y \downarrow \Rightarrow \Delta p_y \uparrow$$
$$\Rightarrow \boxed{\text{spacing} \uparrow}$$

Sample Question

- How do the de Broglie wavelengths of a 100 eV electron and a 1 eV proton compare? The mass of a proton is ~ 1800 times greater than that of an electron. (both are non-relativistic)
 - A. The electron wavelength is greater
 - B. The proton wavelength is greater
 - C. The wavelengths are approximately the same
 - D. Protons don't have a de Broglie wavelength

$$\lambda_0 = h/p = \frac{h}{\sqrt{2mK}}$$

$$\lambda_{0e} / \lambda_{0p} = \frac{h / \sqrt{2m_e \cdot 100 \text{ eV}}}{h / \sqrt{2m_p \cdot 1 \text{ eV}}}$$

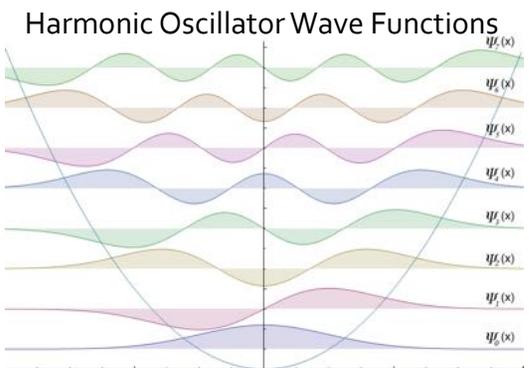
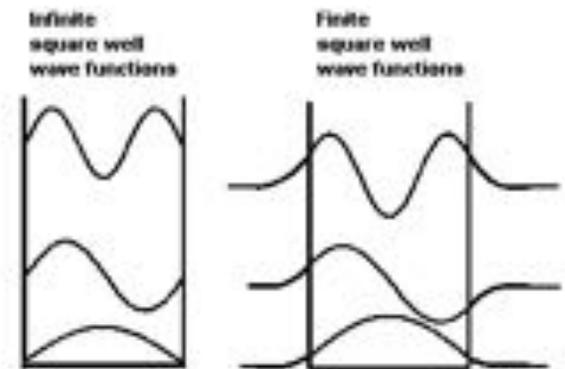
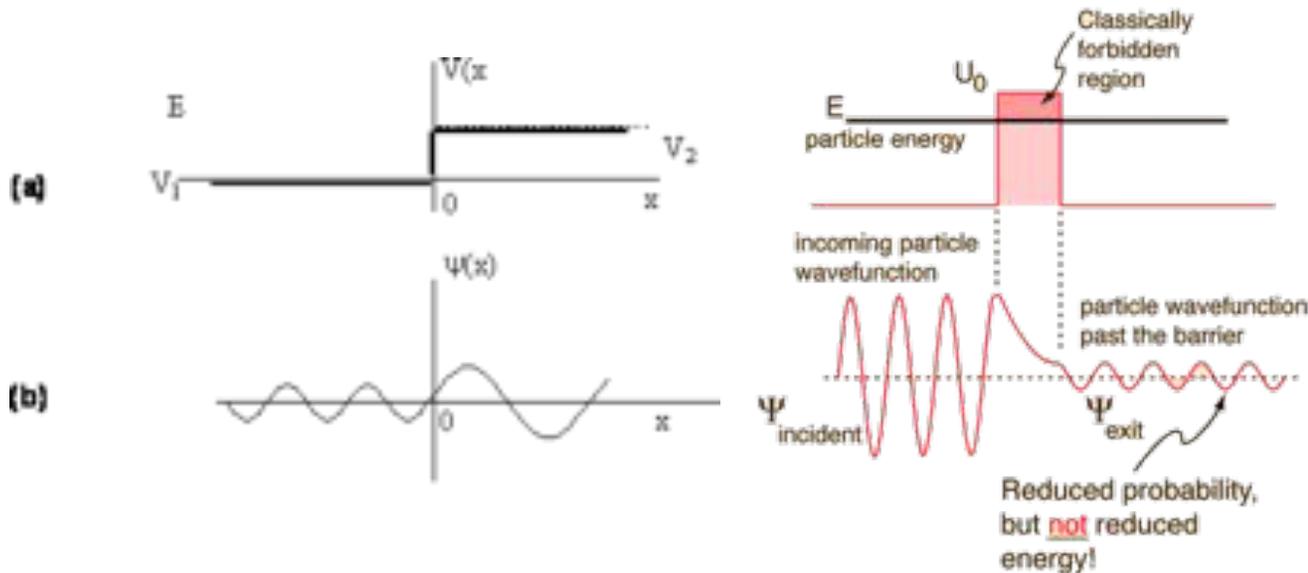
$$= \sqrt{\frac{m_p}{100 m_e}} = \sqrt{\frac{1836}{100}} > 1$$

Schrödinger Equation

$$\frac{-\hbar^2}{2m} \frac{\partial^2 \Psi(x)}{\partial x^2} + U(x)\Psi(x) = E\Psi(x)$$

Time independent equation

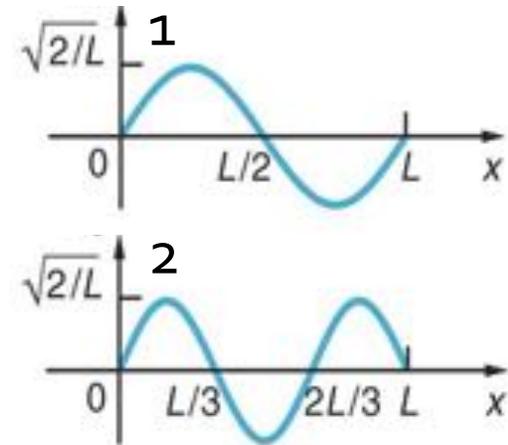
Solutions must be continuous
Continuous in slope unless $U = \infty$



Sample Question

- Two particles are trapped in separate square well potentials (with the same size, depth, etc). The first particle's wave function has one full wavelength across the box. The second particle's wave function has 1.5 wavelengths across the box. Which is in a higher energy level?

- A. First particle
- B. Second particle
- C. Both are the same
- D. No way to tell from information given



$$E = p^2/2m + U$$
$$= \frac{h^2}{2m\lambda^2} + U$$

smaller $\lambda \Rightarrow$ higher kinetic
energy

\Rightarrow higher E for
const. U

Sample Question

- A particle incident on a potential step at $x = 0$ has the wave function $\psi_1 = Ae^{ikx} + Be^{-ikx}$ for $x < 0$, and the wave function $\psi_2 = Ce^{-k'x}$ for $x > 0$. Which equations are correct?
 - A. $A + B = C$
 - B. $ik(A - B) = -k'C$
 - C. $A^2 + B^2 = C^2$
 - D. $A = -B, A = C$
 - E. $A = C, B = 0$

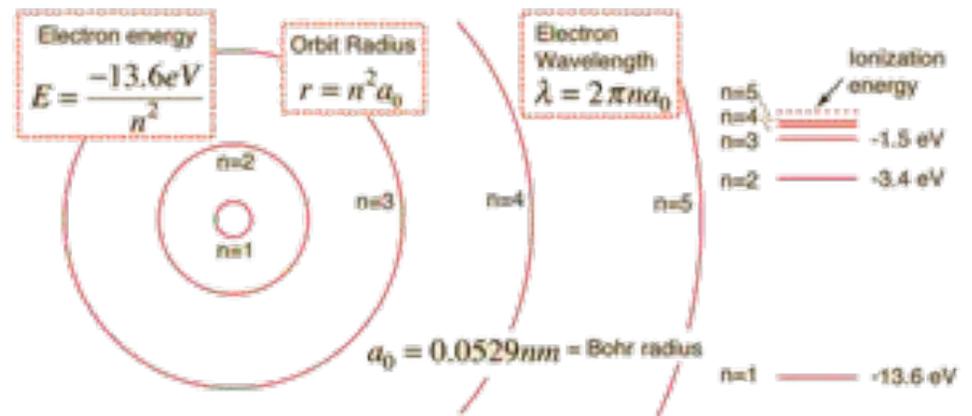
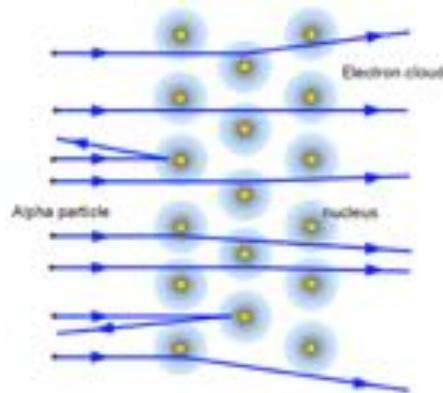
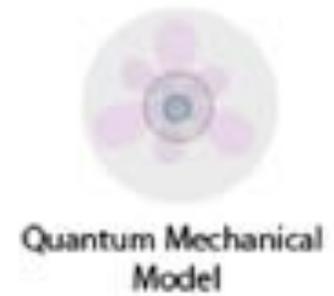
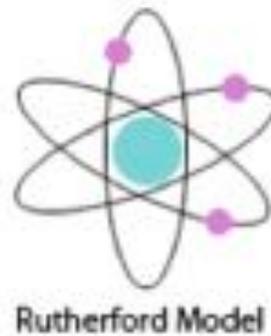
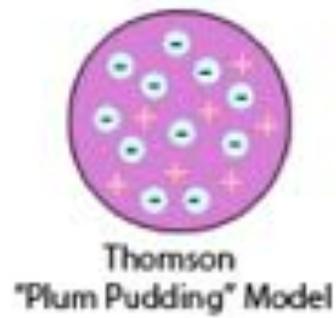
ψ continuous @ $x=0$

$$Ae^0 + Be^0 = Ce^0$$
$$\Rightarrow \boxed{A+B=C}$$

ψ' continuous @ $x=0$

$$ikAe^0 - ikBe^0 = -k'Ce^0$$
$$\Rightarrow \boxed{ik(A-B) = -k'C}$$

Bohr Atom

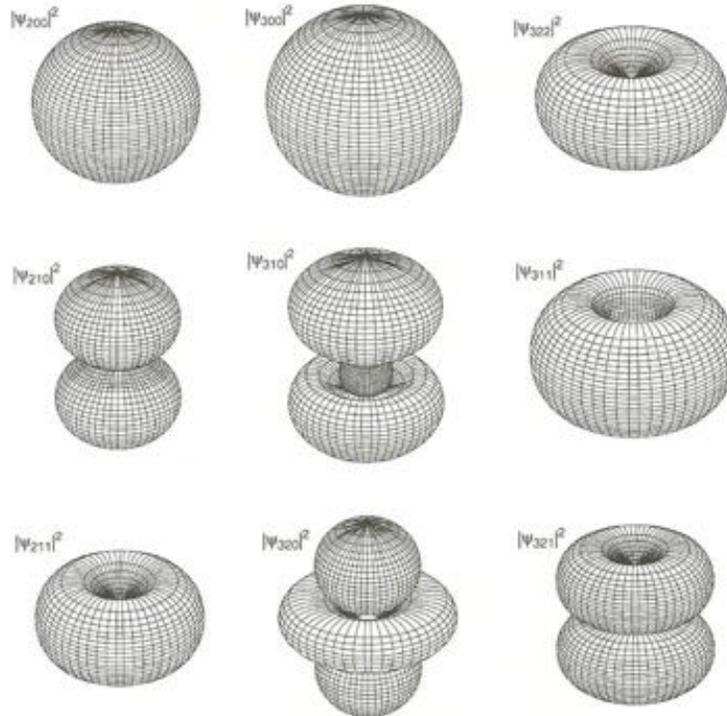
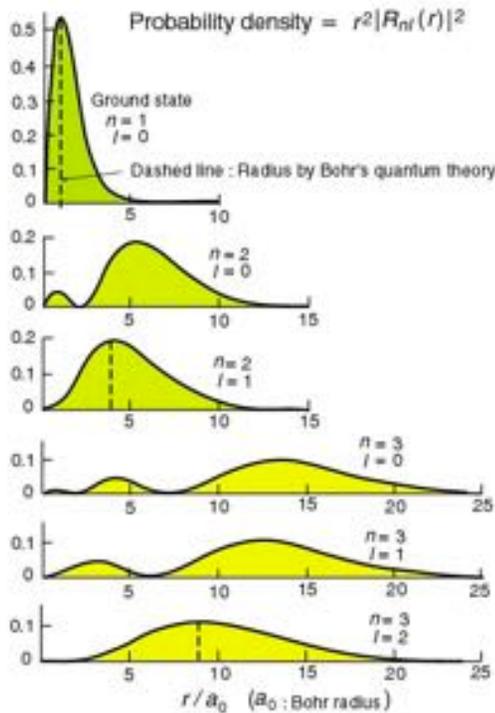


Sample Question

- Imagine an atom of Boron ($Z = 5$) which has been ionized four times, leaving it with a single electron. How much energy would it take to remove that final electron?
- A. 13.6 eV
 - B. $13.6/25 = 0.544$ eV
 - C. $13.6/5 = 2.72$ eV
 - D. $13.6*5 = 68$ eV
 - E. $13.6*25 = 340$ eV

$$\begin{aligned}\Delta E &= 13.6 \frac{z^2}{4^2} \\ &= 13.6 \cdot z^2 \\ &= \boxed{3410 \text{ eV}}\end{aligned}$$

Hydrogen Atom



	n	l	m_l	s
1s	1	0	0	$\frac{1}{2}, -\frac{1}{2}$
2s	2	0	0	$\frac{1}{2}, -\frac{1}{2}$
2p	2	1	1, 0, -1	$\frac{1}{2}, -\frac{1}{2}$
3s	3	0	0	$\frac{1}{2}, -\frac{1}{2}$
3p	3	1	1, 0, -1	$\frac{1}{2}, -\frac{1}{2}$
3d	3	2	2, 1, 0, -1, -2	$\frac{1}{2}, -\frac{1}{2}$
4s	4	0	0	$\frac{1}{2}, -\frac{1}{2}$
4p	4	1	1, 0, -1	$\frac{1}{2}, -\frac{1}{2}$
4d	4	2	2, 1, 0, -1, -2	$\frac{1}{2}, -\frac{1}{2}$
4f	4	3	3, 2, 1, 0, -1, -2, -3	$\frac{1}{2}, -\frac{1}{2}$

n -> Energy
 l -> Angular momentum
 m_l -> Orientation of l
 m_s -> Electron spin

