

College Physics I: 1511

Mechanics & Thermodynamics

Professor Jasper Halekas
Van Allen Lecture Room 1
MWF 8:30-9:20 Lecture

Announcements

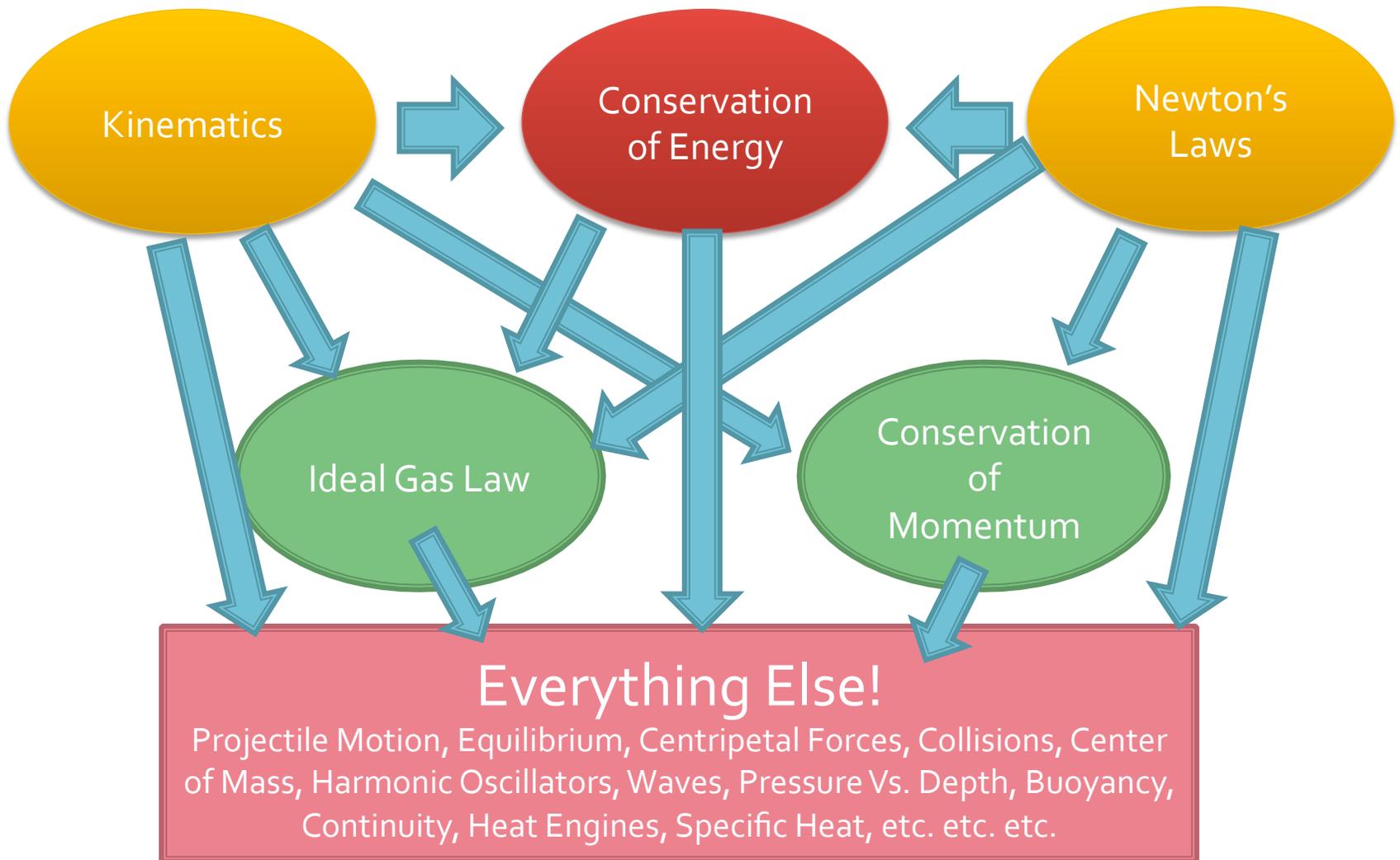
- HW grades in ICON today
- Lab grades in ICON this weekend
- Please fill out evaluations!
 - Completion rate 27% as of 12/9
 - Deadline Saturday 12/10
- Final 8:00-10:00 pm in LR1 on Monday 12/12



The Five Most Important Concepts

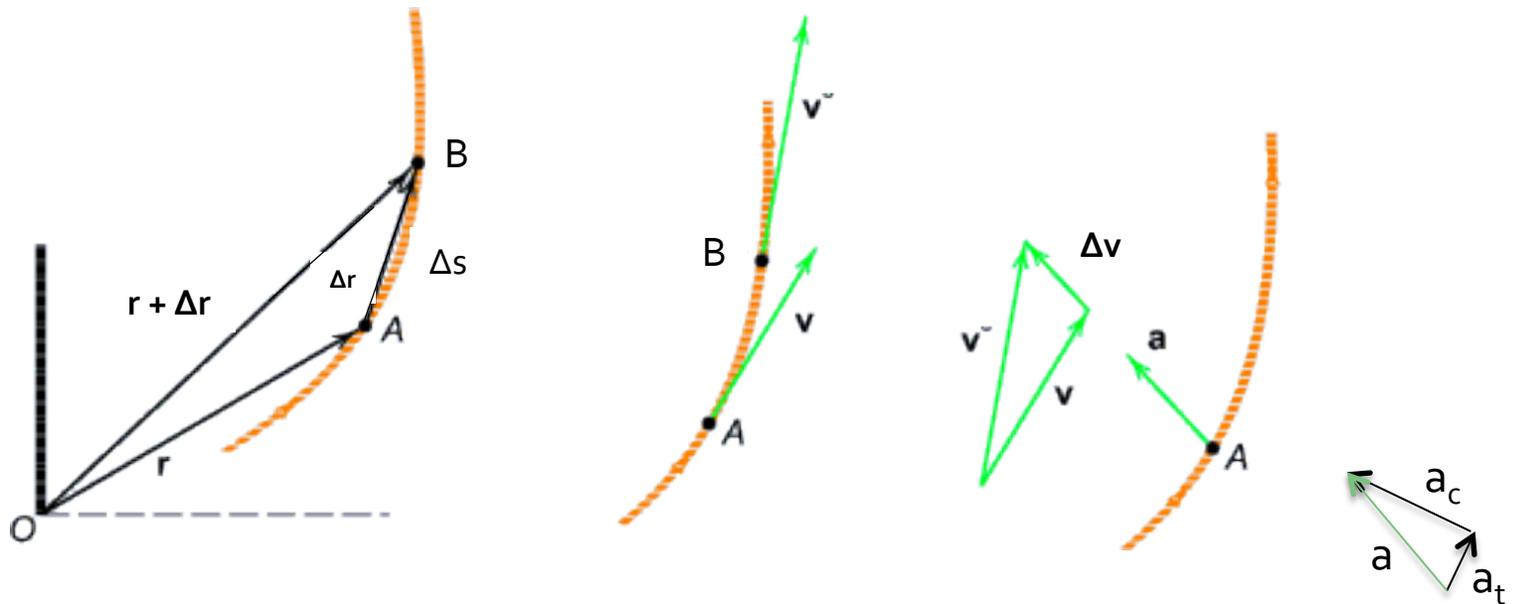
1. Kinematics
 - A. Translational
 - B. Rotational
2. Newton's Laws
3. Conservation of Energy
 - A. Kinetic Energy
 - B. Potential Energy
 - C. Work
 - D. Heat
 - E. Temperature
4. Conservation of Momentum
5. Ideal Gas Law and Processes

Physics Concept Flow Chart



Kinematics is the Start

- Kinematic Variables



$$\langle \vec{v} \rangle = \frac{\Delta \vec{r}}{\Delta t}$$

$$\langle \vec{a} \rangle = \frac{\Delta \vec{v}}{\Delta t}$$

$$\theta = s/r$$

$$\langle \omega \rangle = \frac{\langle v_t \rangle}{r} = \frac{\Delta \theta}{\Delta t}$$

$$\langle \alpha \rangle = \frac{\langle a_t \rangle}{r} = \frac{\Delta \omega}{\Delta t}$$

$$a_c = v^2/r$$

Kinematics is the Start

- Kinematic Equations

Translational	Rotational
$v = v_0 + at$	$\omega = \omega_0 + \alpha t$
$\Delta x = v_0 t + \frac{1}{2} at^2$	$\Delta \theta = \omega_0 t + \frac{1}{2} \alpha t^2$
$v^2 = v_0^2 + 2a\Delta x$	$\omega^2 = \omega_0^2 + 2\alpha\Delta \theta$

Newton's Laws

Newton's Laws

1. A body will remain at rest, or moving at a constant velocity, unless it is acted on by an unbalanced force.
2. The force experienced by an object is proportional to its mass times the acceleration it experiences:

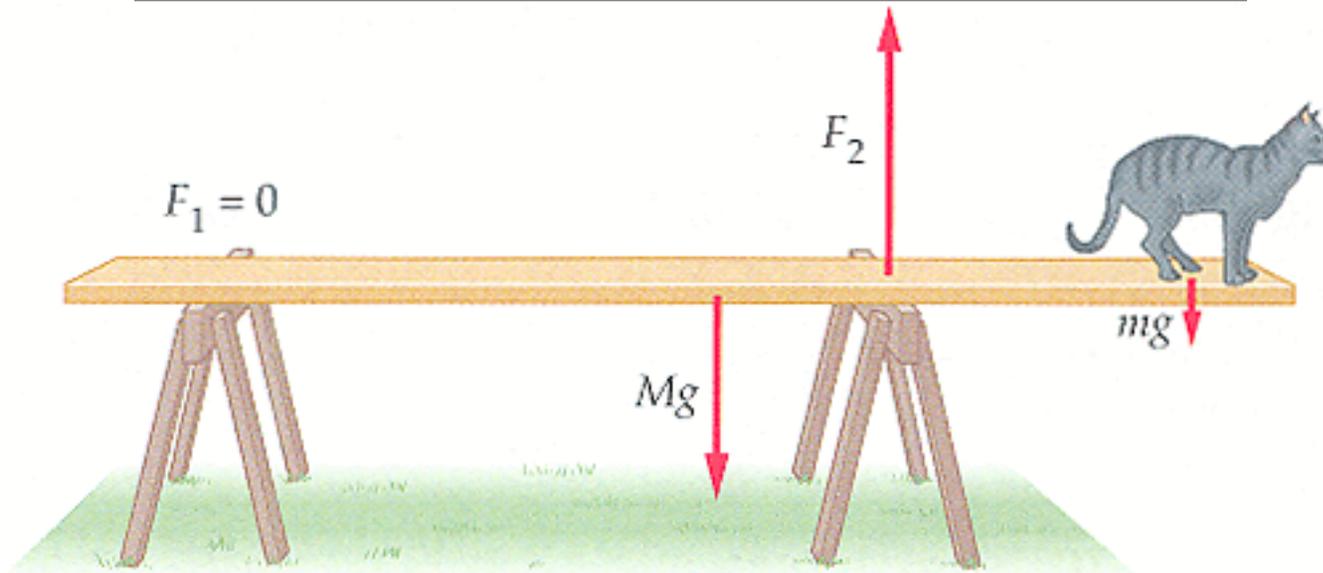
$$\vec{F} = m\vec{a}$$

3. If two bodies exert a force on one another, the forces are equal in magnitude, but opposite in direction:

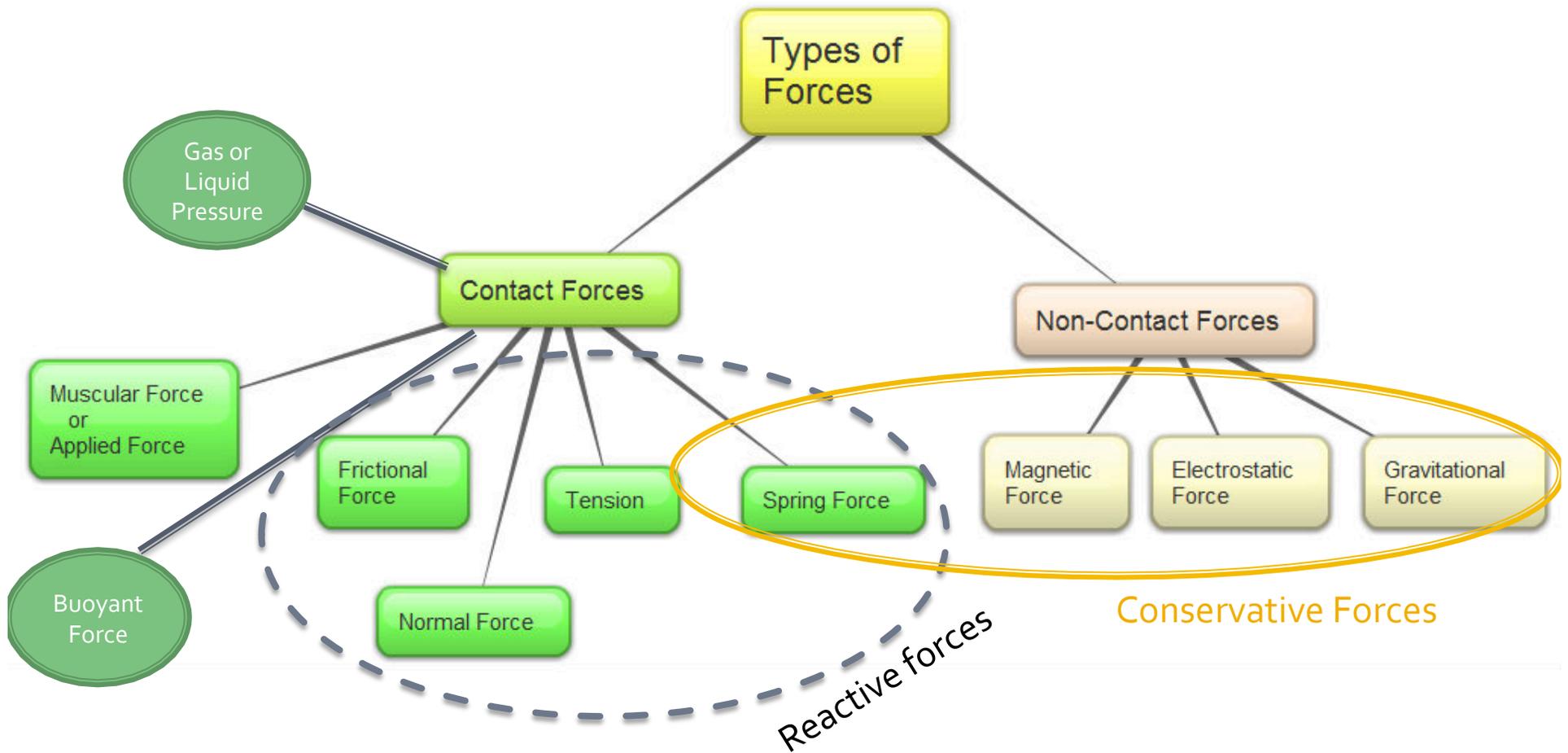
$$\vec{F}_{12} = -\vec{F}_{21}$$

Equilibrium

- Static Equilibrium
 - sum of forces on the body = 0
 - $\Sigma F = 0$
 - sum of torques on the body = 0
 - $\Sigma T = 0$



Forces



Newton + Kinematics -> Work-Energy Theorem

$$W_{net} = F_{net} \Delta x$$

$$W_{net} = ma\Delta x$$

$$v_f^2 = v_i^2 + 2a\Delta x$$

$$a\Delta x = \frac{v_f^2 - v_i^2}{2}$$

$$W_{net} = m\left(\frac{v_f^2 - v_i^2}{2}\right)$$

$$W_{net} = \frac{1}{2}mv_f^2 - \frac{1}{2}mv_i^2$$

$$W_{net} = \Delta KE$$

Also works for rotational kinematics,
but $W = \tau \Delta\theta$ and $KE = \frac{1}{2} I \omega^2$

Work and Potential Energy

- Work done **on** a body by a conservative force comes from the potential energy PE associated with that force:
 - W and ΔPE are equal and opposite if conservative

Gravitational P.E.

$$W_{gravity} = -mg(\Delta y) = -mg(\Delta h)$$

$$PE_g = -W_{gravity} = mgy = mgh$$

$h = 0$ is arbitrarily set by you

E.g. decrease in height

⇒ loss in PE

⇒ positive W by gravity

⇒ increase in KE

Conservation of Mechanical Energy

$$\begin{array}{ccccccccc} \underline{E} & = & \underline{\frac{1}{2}mv^2} & + & \underline{\frac{1}{2}I\omega^2} & + & \underline{mgh} & + & \underline{\frac{1}{2}kx^2} \\ \text{Total} & & \text{Translational} & & \text{Rotational} & & \text{Gravitational} & & \text{Elastic} \\ \text{mechanical} & & \text{kinetic} & & \text{kinetic} & & \text{potential} & & \text{potential} \\ \text{energy} & & \text{energy} & & \text{energy} & & \text{energy} & & \text{energy} \end{array}$$

Mechanical energy can only change if a non-conservative force acts

A “non-conservative” process is code for a process in which some of the mechanical energy is lost to heat or deformation of the body

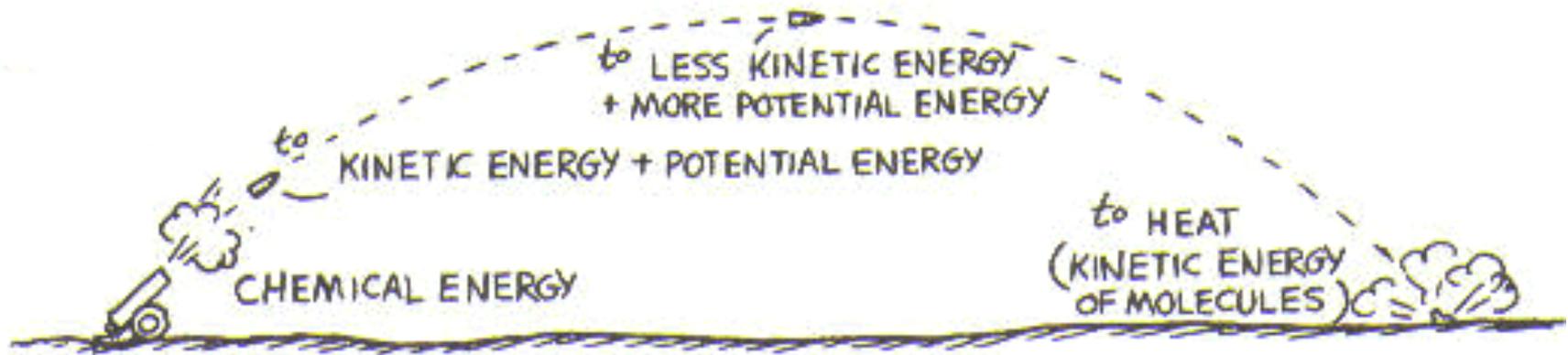
$$\Delta E = W_{nc} \leftarrow \text{Nonconservative work done on body}$$

Conservation of Total Energy

$$\Delta U = Q - W$$

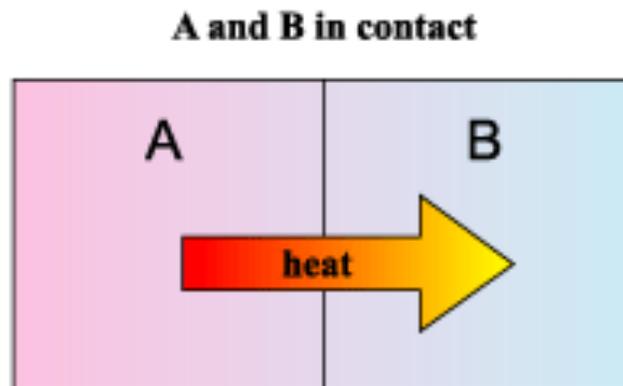
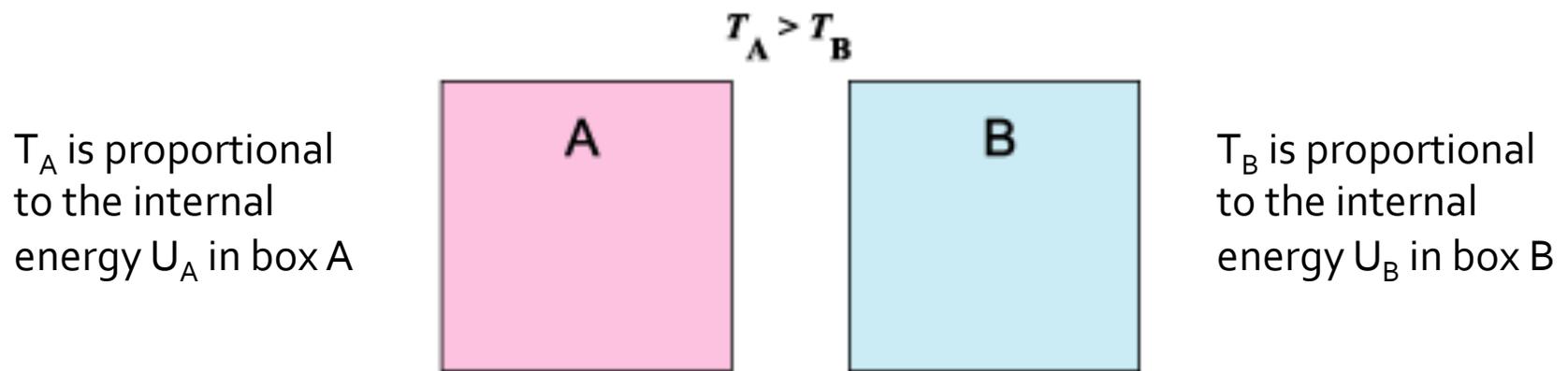
Energy stored or Energy stores Lost Energy added to Energy used by

Nonconservative work done by body



Energy Cannot Be Created or Destroyed
(It just changes forms)

Heat, Internal Energy, Temperature



$$Q = mc\Delta T$$
$$Q = nC\Delta T$$

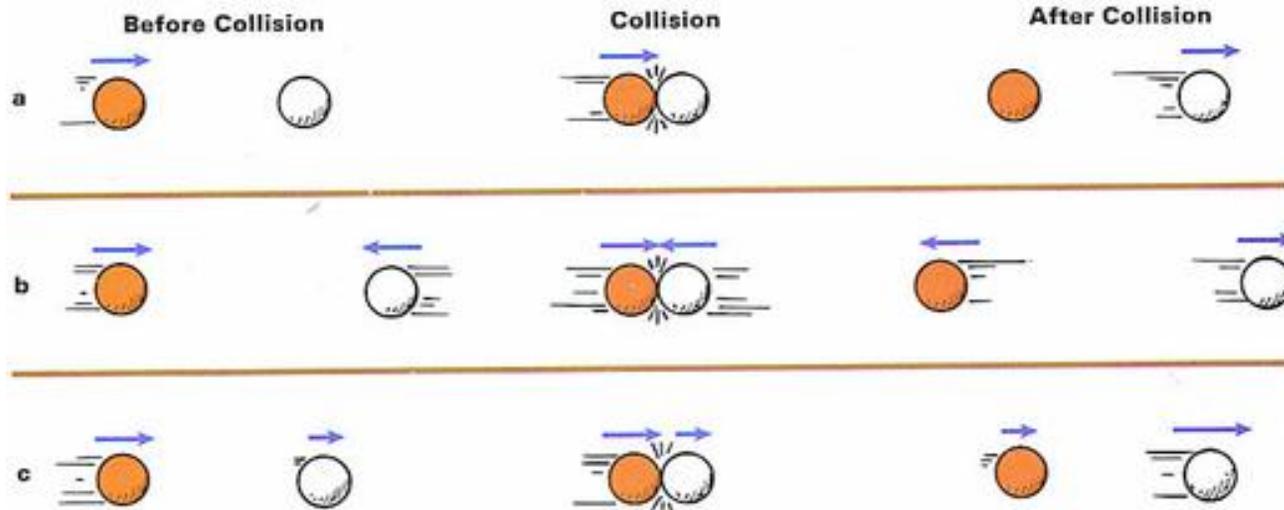
Heat will flow (transferring energy) until $T_A = T_B$

Kinematics + Newton -> Conservation of Momentum

$$p = mv$$

$$F = ma = \frac{m\Delta v}{\Delta t} = \frac{\Delta p}{\Delta t}$$

The average force on a constant mass system is seen to be equal to the rate of change of momentum.



If no external forces :

$$m_1 \vec{v}_{1i} + m_2 \vec{v}_{2i} + \dots + m_n \vec{v}_{ni}$$

$$= m_1 \vec{v}_{1f} + m_2 \vec{v}_{2f} + \dots + m_n \vec{v}_{nf}$$

Ideal Gas Law: An Experimental Fact

Pressure

Temperature

Number of moles

$PV = nRT$

Volume

Gas constant

Pressure (Pa)

Volume (m³)

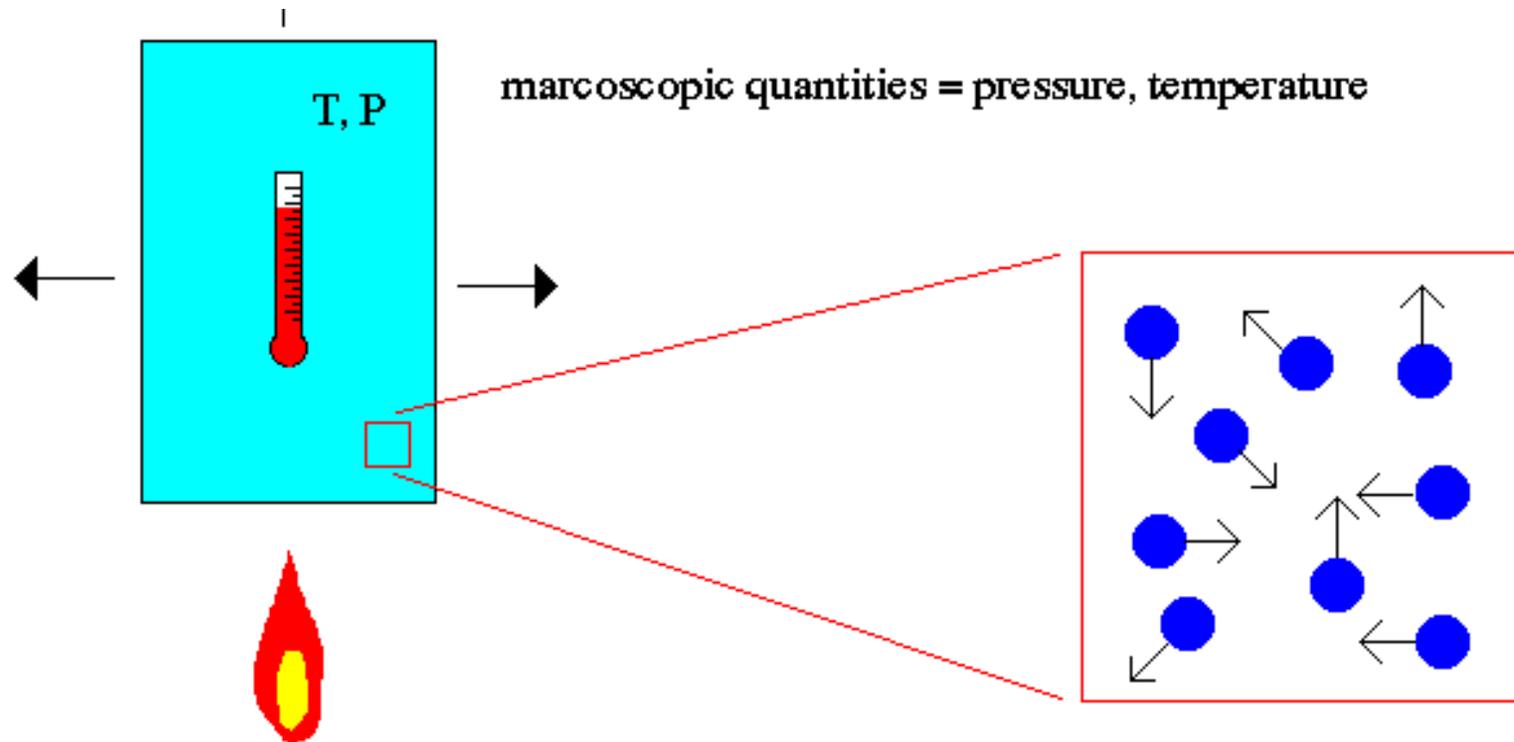
Absolute Temperature (K)

$PV = NkT$

Number of Molecules

Boltzmann's Constant (1.38 x 10⁻²³ J/K)

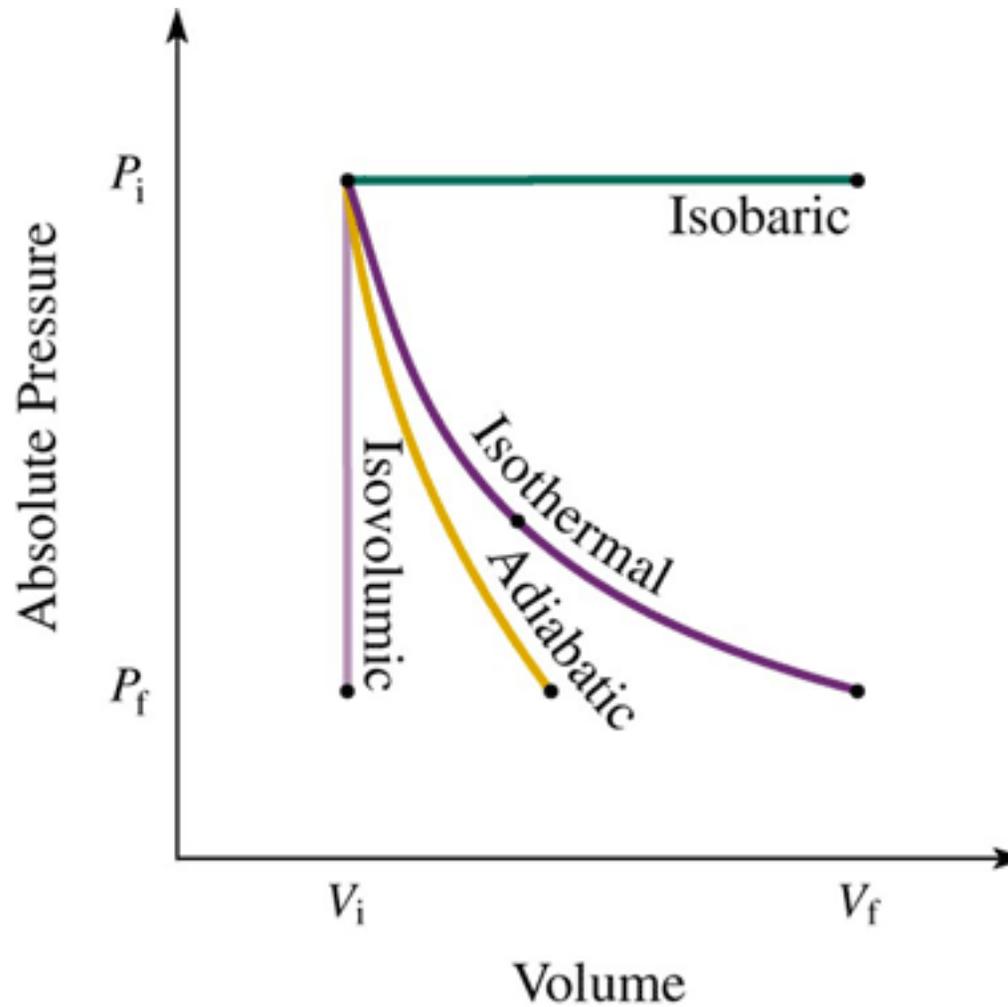
Kinematics + Newton -> Ideal Gas Law



microscopic quantities = kinetic motion of atoms

$$PV = nRT \longleftrightarrow PV = \frac{2}{3} N \left[\frac{1}{2} \overline{mv^2} \right]$$

PV Diagrams

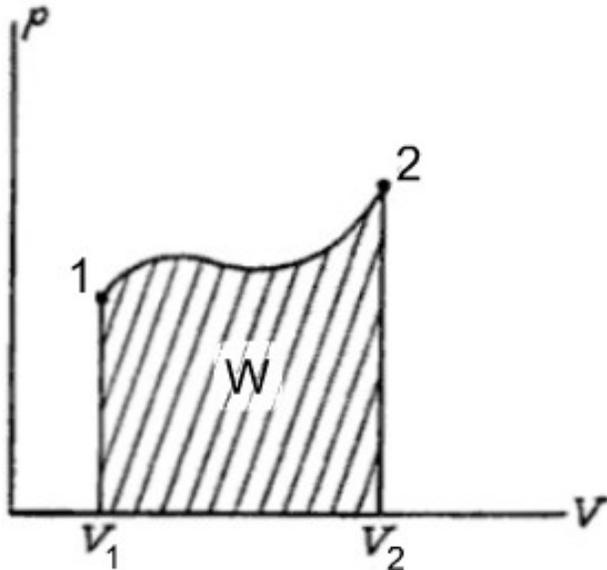


Ideal Gas Processes

ΔU same for all, just depends on ΔT

Process	ΔU	Q	W
Constant Volume	$\frac{3}{2} nR \Delta T$ (monatomic)	$\frac{3}{2} nR \Delta T$ (monatomic)	0 Lower specific heat All Q goes to U
Constant Pressure	$\frac{3}{2} nR \Delta T$ (monatomic)	$\frac{5}{2} nR \Delta T$ (monatomic)	$P\Delta V = nR \Delta T$ Higher specific heat Some Q goes to W
Constant Temperature	0	$nRT \ln(V_f/V_i)$	$nRT \ln(V_f/V_i)$
Adiabatic ($pV^\gamma =$ constant)	$\frac{3}{2} nR \Delta T$ (monatomic)	0	$-\frac{3}{2} nR \Delta T$ (monatomic)

PV Diagrams



Work W done by gas = area under curve

Increase in volume \rightarrow positive work $+W$ by gas

Decrease in volume \rightarrow negative work $-W$ by gas

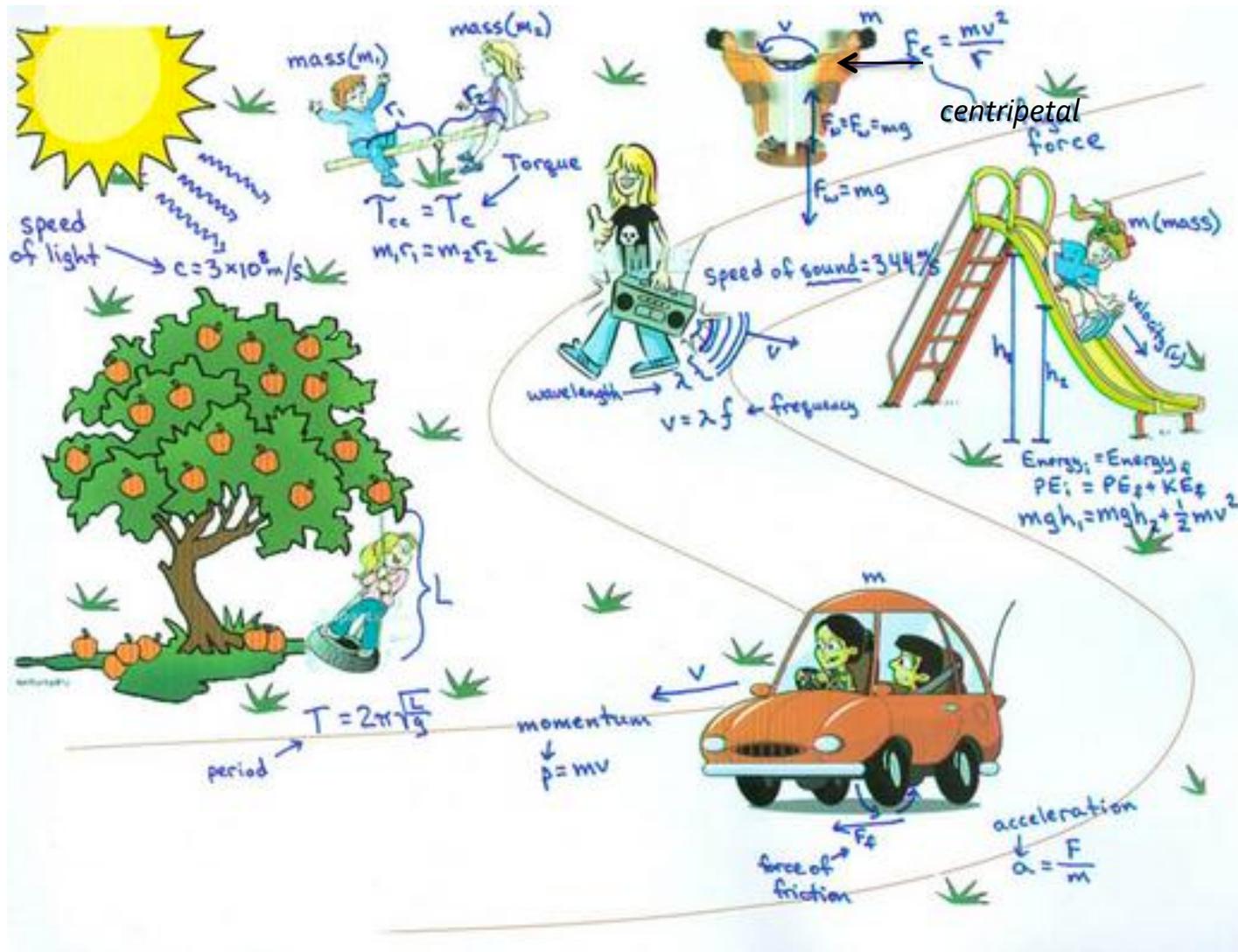
Internal energy U can be read from this diagram as well:

Internal energy U is proportional to temperature T

Temperature T is related to P & V through ideal gas law

From internal energy U and work W , we can get heat Q

Physics in Everyday Life



Conserve Energy



Goodbye!

- You've been a great class!
- You are the future of our world – use your knowledge well
- My door is always open to my former students